

Electrolysis, the Faraday, and Avogadro's Number

1. Final volume in buret 24
2. Initial volume in buret 44.0 ml
3. Volume of hydrogen 20
4. Temperature of solution 23°C
5. Height of water column 33 mm
6. Barometric pressure 740 mm Hg
7. Vapor pressure of H_2O (see Appendix "Answers to Selected Pre-lab Questions") 23.8 mm Hg
8. Pressure of H_2 _____ mm Hg _____ atm (show calculations)

9. Moles of H_2 produced (show calculations) _____

10. Time reaction started 0

11. Time reaction terminated _____

12. Elapsed time 1 min 50 s

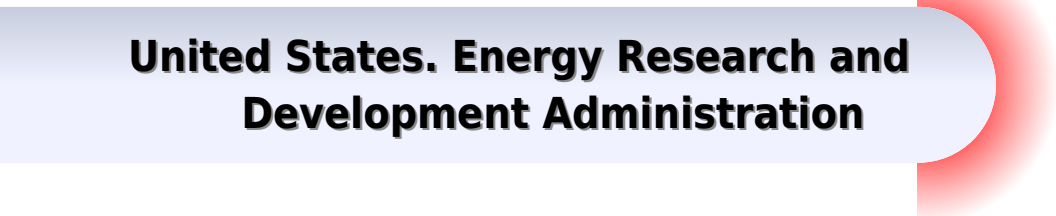
13. Current 1.2 A

14. Number of coulombs passed _____ C

15. Value for faraday _____ Accepted value _____ (show calculations)

Electrolysis Experiment Lab Answer Key

**United States. Energy Research and
Development Administration**



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