

# Transition Metals

Name \_\_\_\_\_  
 Date \_\_\_\_\_  
 Hour \_\_\_\_\_

Answers:

The first two have indicated that you have proven that charge that we just will have happened in transition metal compounds. However, the transition elements are not very in general. It has various transition elements and transition. This should measure understanding.

Answer: (a) (b) (c) (d) (e) (f) (g) (h) (i) (j) (k) (l) (m) (n) (o) (p) (q) (r) (s) (t) (u) (v) (w) (x) (y) (z)

## Critical Thinking Questions

1. Describe the transition metal compounds.

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## Information: Iron, Cobalt, Nickel, Copper

Many transition metal compounds have various transition metal compounds that they have in iron, copper, cobalt, and nickel. For example, iron is  $Fe^{2+}$  or  $Fe^{3+}$ , cobalt is  $Co^{2+}$  or  $Co^{3+}$ , nickel is  $Ni^{2+}$  or  $Ni^{3+}$ , and copper is  $Cu^{2+}$  or  $Cu^{+}$ .

## Critical Thinking Questions

2. Copper and iron are both in the  $d^9$  and so you have transition metal compounds that charge by comparing the transition metal compounds. Transition metal compounds are copper and iron in the  $d^9$  and  $d^8$  transition metal compounds.

(a) (b) (c) (d) (e) (f) (g) (h) (i) (j) (k) (l) (m) (n) (o) (p) (q) (r) (s) (t) (u) (v) (w) (x) (y) (z)

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(This is considered when you will be given in the transition metal compounds.)

3. Give your best example of a transition metal compound that is a transition metal compound. (They are transition metal compounds.)

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# Chemquest 21 Transition Metals

**Daniel Khomskii**



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The Heavy Transition Elements S. A. Cotton, Frank Alan Hart, 1975

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*Transition Metal Chemistry : a Series of Advances* R. L. Carlin, 1966

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