
GCSE Chemistry – Electrolysis Exam Questions and Answers

What is electrolysis? - Answer- passing an electric current through an ionic substance that's molten or in a solution and breaks it down into new substances

What is electrolysis used for? - Answer- To make new substances

What is an electrolyte? - Answer- a liquid that conducts electricity

How are electrolytes made? - Answer- by melting or dissolving ionic compounds

Which part of an electrolyte conducts electricity? - Answer- The free electrons

What is the a positive electrode called? - Answer- Anode

What is a negative electrode called? - Answer- Cathode

What are electrodes made from and why? - Answer- Unreactive metal so they don't take part in the reaction

How do ions become atoms or molecules? - Answer- By losing or gaining electrons

How does an electrolysis circuit become complete? - Answer- A flow of electrons taken from anodes to cathodes.

How do you measure the conductivity of an electrolyte? - Answer- With a conductivity probe

In molten ionic compounds there is only _____ - Answer- one source of ions

Why can molten ionic compounds be electrolysed? - Answer- Because ions can move freely

In the electrolysis of molten lead bromide what is the half equation for the Pb^{2+} ions? - Answer- $\text{Pb}^{2+} + 2\text{e}^{-} \rightarrow \text{Pb}$

In the electrolysis of molten lead bromide what is the half equation for the Br^{-} ions? - Answer- $2\text{Br}^{-} \rightarrow \text{Br}_2 + 2\text{e}^{-}$

Electrolysis Questions And Answers

Kok-Sing Tang, Kristina Danielsson



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Kok-Sing Tang, Kristina Danielsson, 2018-01-19 This book highlights recent developments in literacy research in science teaching and learning from countries such as Australia Brazil China Finland Germany Hong Kong New Zealand Norway Singapore Spain South Africa Sweden Taiwan and the United States It includes multiple topics and perspectives on the role of literacy in enhancing science teaching and learning such as the struggles faced by students in science literacy learning case studies and evaluations of classroom based interventions and the challenges encountered in the science classrooms It offers a critical and comprehensive investigation on numerous emerging themes in the area of literacy and science education including disciplinary literacy scientific literacy classroom discourse multimodality language and representations of science and content and language integrated learning CLIL The diversity of views and research contexts in this volume presents a useful introductory handbook for academics researchers and graduate students working in this specialized niche area With a wealth of instructional ideas and innovations it is also highly relevant for teachers and teacher educators seeking to improve science teaching and learning through the use of literacy

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