Name	10	Date	Class

CHAPTER 8 STUDY GUIDE

Covalent Bonding

Section 8.1 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

covalent bond	molecule	sigma bond	exothermic	pi bond		
	ectrons occurs, the attachr					
a(n) (1)	. When such	an attachment is for	med, bond dissociation	n energy is		
released, and the proce	rss is (2)	, When t	wo or more atoms bone	by means of		
electron sharing, the re	sulting particle is called a	(n) (3)	3) If the electrons shared			
are centered between t	he two atoms, the attachm	ent is called a(n) (4	0	If the		
sharing involves the or	verlap of parallel orbitals,	the attachment is co				
	e choice that best comple dements such as hydrogen			om.		
6. In what form do e	elements such as hydrogen	, nitrogen, and oxy	gen normally occur?			
	a. as single atoms		 as molecules containing three atoms 			
 b. as molecules c 	containing two atoms	d. as mol	 as molecules containing four atoms 			
7. How many electro	ons are shared in a double	covalent bond?				
a. none	b. one	c. two	d. f	our		
8. Bond length is the	e distance between					
a. two molecules	s of the same substance.	c. the nu	elei of two attached ato	ms.		
 b. the electrons i 	n two attached atoms.	d. the orb	itals of two attached at	oms.		
9. Which of the follo	Which of the following relationships relating to bond length is generally correct?					
a. the shorter the	bond, the stronger the bo	nd				
	bond, the weaker the bon					
	bond, the fewer the elect					
 d. the shorter the 	bond, the lower the bond	dissociation energy	y			

Covalent Bonding Study Guide Key

Judd E. Hollander

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