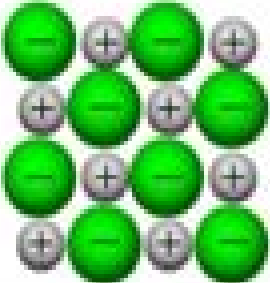
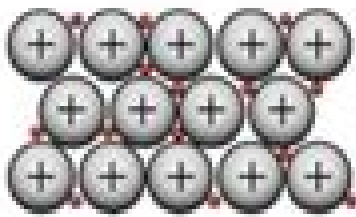
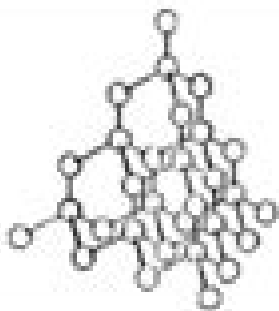
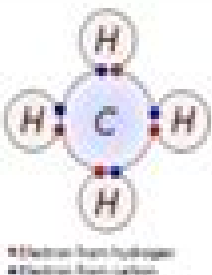


Metallic Bonding	Giant Ionic Structure	Simple covalent structure	Giant covalent structure
			
Positive ions surrounded by a 'sea' of delocalised electrons	Regular 3D arrangement of positive and negative ions	Isolated molecules with weak forces between molecules	Repeating 3D structure with strong bonds between atoms
Strong attraction between ions and delocalised electrons	Strong electrostatic attraction between positive and negative ions	Strong bonds between atoms, electrons are shared	Strong bonds between atoms, electrons are shared
High melting points, almost all are solids at room temp	High melting points, always solid at room temp	Low melting points, gases and liquids at room temp	High melting points, always solid at room temp
Conduct electricity	Don't conduct electricity	Diamond	Conduct heat
Most don't conduct electricity (except graphite)	Conduct electricity when molten and in solution	Regular crystal arrangement	Regular crystal arrangement
Sodium chloride	Iodine	Sodium	Malleable

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