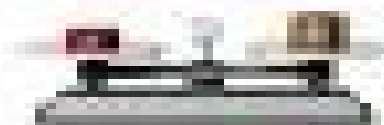


## Chapter 3: Atoms and

### Activity 1: Balancing a chemical equation

Write a balanced chemical equation for the reaction between hydrogen and oxygen to form water.



	Red weight	Brown weight	Number of atoms	Relative atomic mass
Left pan	1	0	1	1
Right pan	0	1	1	16
Balance	1	1	1	16

16g of oxygen reacts with 1g of hydrogen to form 17g of water.

### Activity 2: Balancing a chemical equation

Write a balanced chemical equation for the reaction between hydrogen and oxygen to form water.



	Brown weight	Blue weight	Number of atoms	Relative atomic mass
Left pan	1	0	1	16
Right pan	0	1	1	16
Balance	1	1	1	16

16g of oxygen reacts with 1g of hydrogen to form 17g of water.

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textbooks on the subject and emphasises the two important topics underpinning the subject quantum mechanics and the second law of thermodynamics Both topics are challenging to students because they focus on uncertainty and probability The book explains these fundamental concepts clearly and shows how they offer the key to understanding a wide range of chemical phenomena including atomic and molecular spectra the structure and properties of solids liquids and gases chemical equilibrium and the rates of chemical reactions This revised edition has enabled improvements and corrections to be made

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