

Chemistry Study Guide – Stoichiometry

Name _____

Date _____

Period _____

1 mole is defined as _____ particles, also known as Avogadro's number.

Molar Mass – Calculate for the molar mass of each compound



Balancing Equations – Balance each of the following equations



Simple Stoichiometry

- How many atoms of Na are present in a 34.5 g sample of Na metal?
- How many atoms of K are present in a 58.8 g sample of K metal?
- How many liters of O_2 are present in a 60.4 g sample of O_2 gas?
- How many liters of NH_3 are present in a 105.3 g sample of NH_3 gas?
- How many grams of HCl are present in a sample of 5.04×10^{24} atoms HCl?
- How many grams of H_2 are present in a 13.2 L sample of H_2 gas?

Chemistry Study Guide Stoichiometry

Christian G. Meyer



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