

10.1 Periodic Trends



10.1 PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

Section 10.1: Introduction

In your notebook, read aloud the modern periodic table.

Copy the letter numbers in the boxes on the left below from the periodic table to complete the table on the right.

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 10. It exists in the periodic table.
- _____ 11. It has no line periodic table.
- _____ 12. Group 1 elements.
- _____ 13. Elements that are strong oxidizing agents.
- _____ 14. Group 18 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write each of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- _____ 15. There are two representative elements of elements.
- _____ 16. Most elements have a line of the periodic table.
- _____ 17. Group 1 elements are highly reactive and form in the alkali metals.
- _____ 18. Group 18 elements are the noble gases.
- _____ 19. Group 1 elements are highly reactive elements in the periodic table.
- _____ 20. Group 18 elements are very reactive elements in the periodic table.
- _____ 21. Elements have properties of both metals and nonmetals.

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Hussin A.Rothana



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