

CHAPTER 8

STUDY GUIDE

Covalent Bonding

Section 8.1 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

covalent bond

molecule

sigma bond

exothermic

pi bond

When sharing of electrons occurs, the attachment between atoms that results is called a(n) (1) _____. When such an attachment is formed, bond dissociation energy is released, and the process is (2) _____. When two or more atoms bond by means of electron sharing, the resulting particle is called a(n) (3) _____. If the electrons shared are centered between the two atoms, the attachment is called a(n) (4) _____. If the sharing involves the overlap of parallel orbitals, the attachment is called a(n) (5) _____.

In your textbook, read about single and multiple bonds and bond strength.

Circle the letter of the choice that best completes the statement or answers the question.

6. In what form do elements such as hydrogen, nitrogen, and oxygen normally occur?
 - a. as single atoms
 - b. as molecules containing two atoms
 - c. as molecules containing three atoms
 - d. as molecules containing four atoms
7. How many electrons are shared in a double covalent bond?
 - a. none
 - b. one
 - c. two
 - d. four
8. Bond length is the distance between
 - a. two molecules of the same substance.
 - b. the electrons in two attached atoms.
 - c. the nuclei of two attached atoms.
 - d. the orbitals of two attached atoms.
9. Which of the following relationships relating to bond length is generally correct?
 - a. the shorter the bond, the stronger the bond
 - b. the shorter the bond, the weaker the bond
 - c. the shorter the bond, the fewer the electrons in it
 - d. the shorter the bond, the lower the bond dissociation energy

Chemistry Study Guide Covalent Bonding

Yan Bai

A decorative red circular graphic with a gradient, appearing as a partial circle or a stylized 'C' shape, located to the right of the author's name.

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Table of Contents Chemistry Study Guide Covalent Bonding

1. Understanding the eBook Chemistry Study Guide Covalent Bonding
 - The Rise of Digital Reading Chemistry Study Guide Covalent Bonding
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Study Guide Covalent Bonding
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Study Guide Covalent Bonding
 - User-Friendly Interface

4. Exploring eBook Recommendations from Chemistry Study Guide Covalent Bonding
 - Personalized Recommendations
 - Chemistry Study Guide Covalent Bonding User Reviews and Ratings
 - Chemistry Study Guide Covalent Bonding and Bestseller Lists
5. Accessing Chemistry Study Guide Covalent Bonding Free and Paid eBooks
 - Chemistry Study Guide Covalent Bonding Public Domain eBooks
 - Chemistry Study Guide Covalent Bonding eBook Subscription Services
 - Chemistry Study Guide Covalent Bonding Budget-Friendly Options
6. Navigating Chemistry Study Guide Covalent Bonding eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Study Guide Covalent Bonding Compatibility with Devices
 - Chemistry Study Guide Covalent Bonding Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Study Guide Covalent Bonding
 - Highlighting and Note-Taking Chemistry Study Guide Covalent Bonding
 - Interactive Elements Chemistry Study Guide Covalent Bonding
8. Staying Engaged with Chemistry Study Guide Covalent Bonding
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Study Guide Covalent Bonding
9. Balancing eBooks and Physical Books Chemistry Study Guide Covalent Bonding
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Study Guide Covalent Bonding
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Study Guide Covalent Bonding
 - Setting Reading Goals Chemistry Study Guide Covalent Bonding
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Chemistry Study Guide Covalent Bonding
 - Fact-Checking eBook Content of Chemistry Study Guide Covalent Bonding
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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