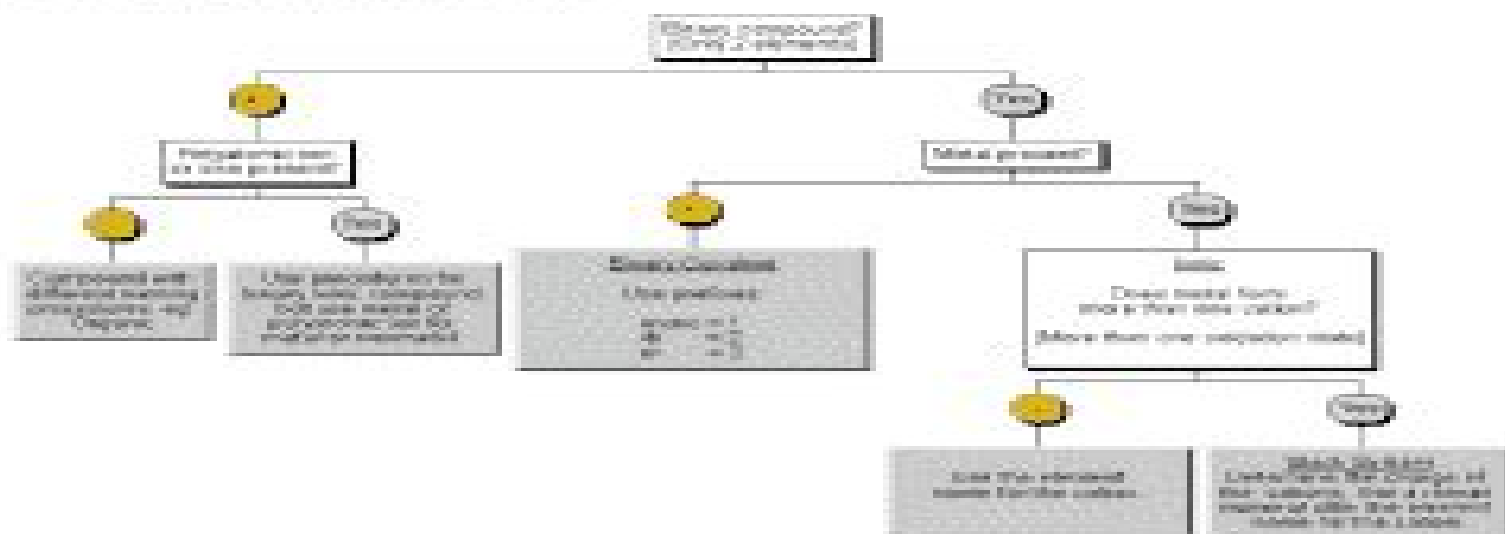


Naming: Putting it all Together

Below is a flow chart showing the steps to follow when naming a compound from the formula. Following the flow chart will enable you to determine when to use the stock system, when to use a set of prefixes, and when to look up the names of polyatomic ions in Table K.



The compounds below are of several different types. Use the flow chart to determine the naming system to use and name each compound shown below.

- | | |
|---|---|
| 1. $\text{Fe}(\text{NO}_3)_3$ | 9. Br_2O |
| 2. $\text{Na}_2\text{S}_2\text{O}_3$ | 10. Hg_2SO_4 |
| 3. P_2O_5 | 11. Al_2O_3 |
| 4. BaBr_2 | 12. SCL_2 |
| 5. $\text{Mn}_2(\text{Cr}_2\text{O}_7)_4$ | 13. IF_7 |
| 6. CaCl_2 | 14. $\text{CH}_3\text{CO}_2\text{Li}$ |
| 7. $(\text{NH}_4)_2\text{S}$ | 15. KNO_3 |
| 8. CaF_2 | 16. AsF_3 |

Evan Chemistry Corner Answer Keys

B Lingard



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