1 This question is about reversible reactions.

Sulfur trioxide is produced in a reversible reaction.

The equation for the reaction is:

sulfur dioxide + oxygen ⇒ sulfur trioxide

1.1 What is the correct balanced equation for the reaction?

Tick one box.

[1 mark]

1.2 The reaction of sulfur dioxide and oxygen to form sulfur trioxide is exothermic.

Which of the following statements is true?

Tick one box.

	The activation	energy	for the	reaction	of	sulfur	dioxide	with	oxygen is	negative.
--	----------------	--------	---------	----------	----	--------	---------	------	-----------	-----------

Sulfur trioxide is colder than sulfur dioxide and oxygen.

Heating the reaction will cause more sulfur trioxide to form.

The reaction of sulfur trioxide to form sulfur dioxide and oxygen is endothermic.

[1 mark]

Gcse Chemistry Practice Papers Higher

M Walker

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