

Chemical Bonding

A. Electronegativity

The ability of an atom to attract electrons in a chemical bond is called electronegativity. The Pauling electronegativity scale is the most commonly used scale for electronegativity. It ranges from 0 to 4.0.



1. Draw a line through the periodic table showing the trend of electronegativity. How does electronegativity change across the periodic table?

Electronegativity increases from left to right and from bottom to top.

2. Why is Fluorine the most electronegative element?

Fluorine has the highest electronegativity because it has the highest effective nuclear charge and the smallest atomic radius.

3. How does electronegativity change across the periodic table?

Electronegativity increases from left to right and from bottom to top. It is highest for Fluorine and lowest for Francium.



Dc Heath And Company Biology Worksheet Answers

Michael Hoelscher



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