



Unit: Gas Laws "Boyle's and Charles's Laws" - Wksh #2



Directions: Please show all work and round each answer the correct number of significant figures. ☺

1. A gas occupies 800.0 mL at a pressure of 780 mm Hg. What is the volume of the gas when its pressure is increased to 880 mm Hg?

2. A gas has a pressure of 0.450 atm and a volume of 32 L. At what pressure would the volume of the gas change to 28L?

3. A gas occupies 525 mL at a pressure of 85.0 kPa. What would the volume of the gas be at a pressure of 65.0 kPa?

4. A gas occupies 3.5 L at standard pressure. Find the volume of the gas when the pressure is 1140 mm Hg.

5. A gas has a volume of 750 mL at a pressure of 2.15 atm. What will the pressure be if the volume becomes 1.25 L?

Gas Laws Homework Answers

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