



QUESTIONS

14. Answer these questions.

- 14.1 Which experiment would provide the most accurate measurement of the mass of a sample of sodium chloride solution?

- A. Weigh 10 g of NaCl.
- B. Weigh 10 g of NaCl, add 100 mL of water.
- C. Weigh 10 g of NaCl, add 100 mL of water, and then weigh the solution.
- D. Weigh 10 g of NaCl, add 100 mL of water, and then weigh the solution.

- 14.2 The following experiment was used to determine the mass of a sample of sodium chloride solution. The mass of the sample was determined by weighing the sample in a beaker. The mass of the sample was determined by weighing the sample in a beaker. The mass of the sample was determined by weighing the sample in a beaker. The mass of the sample was determined by weighing the sample in a beaker.

Mass of sample (g)	Mass of beaker (g)	Mass of solution (g)
10.0	100.0	110.0
20.0	100.0	120.0
30.0	100.0	130.0
40.0	100.0	140.0
50.0	100.0	150.0

- 14.3 Which of the following is the most accurate measurement of the mass of a sample of sodium chloride solution?

- A. 10.0 g
- B. 10.0 g
- C. 10.0 g
- D. 10.0 g

- E. 10.0 g
- F. 10.0 g

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