
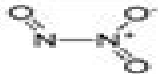


Lewis Dot Structure	Chemical Formula	Name
$\text{:C}\equiv\text{O:}$	CO	carbon monoxide
$\text{:}\ddot{\text{O}}=\text{C}=\ddot{\text{O}}\text{:}$	CO ₂	carbon dioxide
	BF ₃	boron trifluoride
$\text{:N}\equiv\text{N}-\ddot{\text{O}}\text{:}$	N ₂ O	dinitrogen <u>monoxide</u>
	N ₂ O ₃	dinitrogen trioxide

CO₂



2 is a **subscript number**. It appears slightly below the regular line and stands for the number of atoms of the element that comes before it in a compound. If there is not subscript, it means there is only one element.

Covalent Compounds Study Guide

J Ma



Covalent Compounds Study Guide:

Decoding **Covalent Compounds Study Guide**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Covalent Compounds Study Guide**," a mesmerizing literary creation penned with a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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