

Student Name: _____

Class Period: _____

Counting Atoms Practice

<p>1. PH_3</p> <p>Atoms of P: _____</p> <p>Atoms of H: _____</p>	<p>2. HCl</p> <p>Atoms of H: _____</p> <p>Atoms of Cl: _____</p>	<p>3. KNO</p> <p>Atoms of K: _____</p> <p>Atoms of N: _____</p> <p>Atoms of O: _____</p>
<p>4. SiO_2</p> <p>Atoms of Si: _____</p> <p>Atoms of O: _____</p>	<p>5. CaCO_3</p> <p>Atoms of Ca: _____</p> <p>Atoms of C: _____</p> <p>Atoms of O: _____</p>	<p>6. $\text{Na}_2\text{S}_2\text{O}_3$</p> <p>Atoms of Na: _____</p> <p>Atoms of S: _____</p> <p>Atoms of O: _____</p>
<p>7. MgCl_2</p> <p>Atoms of Mg: _____</p> <p>Atoms of Cl: _____</p>	<p>8. $\text{C}_{12}\text{H}_{22}\text{O}_{11}$</p> <p>Atoms of C: _____</p> <p>Atoms of H: _____</p> <p>Atoms of O: _____</p>	<p>9. $\text{KHC}_4\text{H}_4\text{O}_4$</p> <p>Atoms of K: _____</p> <p>Atoms of H: _____</p> <p>Atoms of C: _____</p> <p>Atoms of O: _____</p>
<p>10. CF_2Cl_2</p> <p>Atoms of C: _____</p> <p>Atoms of F: _____</p> <p>Atoms of Cl: _____</p>	<p>11. CH_3OH</p> <p>Atoms of C: _____</p> <p>Atoms of H: _____</p> <p>Atoms of O: _____</p>	<p>12. Na_2HCO_3</p> <p>Atoms of Na: _____</p> <p>Atoms of H: _____</p> <p>Atoms of C: _____</p> <p>Atoms of O: _____</p>
<p>13. $\text{Mg}(\text{OH})_2$</p> <p>Atoms of Mg: _____</p> <p>Atoms of O: _____</p> <p>Atoms of H: _____</p>	<p>14. $\text{Al}_2(\text{SO}_4)_3$</p> <p>Atoms of Al: _____</p> <p>Atoms of S: _____</p> <p>Atoms of O: _____</p>	<p>15. $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$</p> <p>Atoms of Ca: _____</p> <p>Atoms of S: _____</p> <p>Atoms of O: _____</p> <p>Atoms of H: _____</p>

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