

Quiz

Section: Describing Chemical Reactions

In the space provided, write the letter of the term or phrase that best answers this question.

1. The formation of a gas is evidence of
 - a. a chemical change.
 - b. a physical change.
 - c. an change in energy.
 - d. both (a) and (c).
2. Which of the following would be evidence of a chemical change?
 - a. evolution of energy as heat.
 - b. production of a new substance.
 - c. production of a precipitate.
 - d. All of the above.
3. In the word equation,

$$\text{iron} + \text{copper(II) sulfate} \rightarrow \text{copper} + \text{iron(II) sulfate} + \text{energy}$$
 a precipitate is
 - a. iron(II) sulfate.
 - b. copper.
 - c. iron.
 - d. both (a) and (b).
4. In this reaction identified by the following equation,

$$\text{copper} + \text{energy} \rightarrow \text{hydrogen gas} + \text{copper gas}$$
 energy
 - a. can be considered a reactant.
 - b. is released.
 - c. is released.
 - d. both (a) and (b).
5. In this unbalanced chemical equation,

$$\text{copper} + \text{hydrogen} \rightarrow \text{copper}, \text{ the formation of } \text{Cu}_2\text{O} \text{ represents}$$
 - a. a reactant and product.
 - b. a reactant and product.
 - c. a reactant and product.
 - d. a product and product.

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RD Boyd



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