



Database Systems

MODELS, LANGUAGES, DESIGN, AND
APPLICATION PROGRAMMING

Sixth Edition

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$$= \frac{1000 \times \text{molarity}}{1000 + \text{molarity} \times \text{molar mass of solute}}$$

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$$m_{mix} = \frac{m_1 v_1 + m_2 v_2 + \dots + m_n v_n}{V_{total}}$$

m mix m 1 v 1 m 2 v 2 v 1 v 2 1 5 x 480 1 2 x 520 480 520 720 624 1000 1 344 m q5 with increase in temperature which of these changes a molality b fraction of solute present in water

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