

Quarter 3 Exam Review

Multiple Choice

Always, the choice that best completes the statement or answers the question.

- _____ 1. In a chemical reaction
- the mass of the reactants equals the mass of the products.
 - the mass of the product is greater than the mass of reactants.
 - the material atoms in the reactants and products must change.
 - energy is lost must be added in the reaction.
- _____ 2. When a ratio, provided by a chemical reaction, compares facts the solution it is called
- a percent.
 - a percent.
 - a percent.
 - the mass of the product.
- _____ 3. In writing a chemical equation that produces hydrogen gas, the correct representation of hydrogen gas is
- H
 - H₂
 - H₂
 - H₂
- _____ 4. To balance a chemical equation, it may be necessary to adjust the
- coefficients.
 - formulas of the products.
 - subscripts.
 - reactant of products.
- _____ 5. According to the law of conservation of mass, the total mass of the reacting substances is
- always less than the total mass of the products.
 - always less than the total mass of the products.
 - sometimes more and sometimes less than the total mass of the products.
 - always equal to the total mass of the products.
- _____ 6. In the chemical equation $2\text{Mg}(s) + \text{O}_2(g) \rightarrow 2\text{MgO}(s)$
- Mg represents the product magnesium.
 - Mg represents the reactant magnesium.
 - the reaction yield magnesium.
 - O₂ represents the product oxygen gas.
- _____ 7. When the reaction $2\text{FeCl}_3 + 3\text{H}_2\text{O} \rightarrow \text{Fe}_2\text{O}_3 + 6\text{HCl}$ is correctly balanced, what is the coefficient of Fe?
- 4
 - 2
 - 3
 - 6
- _____ 8. Which coefficients correctly balance the formula equation $\text{CaCl}_2 + \text{H}_2\text{O} \rightarrow \text{Ca}(\text{OH})_2$?
- 1, 1, 2
 - 1, 2, 1
 - 1, 1, 1
 - 1, 2, 1
- _____ 9. In which kind of reaction do two or more substances combine to form a new compound?
- decomposition reaction
 - double displacement reaction
 - single reaction
 - synthesis reaction
- _____ 10. The equation $\text{A} + \text{B} \rightarrow \text{C} + \text{D}$ is the general equation for a
- synthesis reaction.
 - decomposition reaction.
 - single displacement reaction.
 - double displacement reaction.

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hybrid cross they contain two genes for each trait