



ANSWER KEY

How many atoms of **oxygen (O)** are in the **chemical formula** below?

- A 4
- B 5
- C 7
- D 12



(d)

The diagram shown below is a **structural formula** for the gas **propane**. What would be the correct **chemical formula** for propane?

- A C_3H_8
- B CH_4
- C $8\text{H}_3\text{C}$
- D CH



(a)

Carbon (C) and hydrogen (H) create **covalent bonds** to share **electrons** between them. What is the total number of electrons being shared between **carbon and hydrogen** in the molecule of propane shown below?

- A 8
- B 11
- C 16
- D 32



(c)

The **reason** why the bonds between carbon (C) and hydrogen (H) are **covalent** is _____

- A C and H do not have electrons to give away
- B C and H are both nonmetals
- C C and H are both metalloids
- D C and H can bond only by sharing electrons



(d)

The diagram below represents the **structural formula** for several **water molecules**. How many **molecules** are represented?

- A 3
- B 5
- C 10
- D 15



(b)

Explain the **arrangement** of the molecules in the diagram of water below.

- A It is a lucky arrangement
- B water molecules are sticky
- C water molecules have polarity
- D water molecules are magnetic



(c)

Water molecules have some **polarity** in their structure. Using the diagram below, notice how the ends of H atoms (**positive**) and O atoms (**negative**) meet. This property is the **reason** that _____

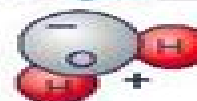
- A water forms droplets easily
- B water boils easily
- C ice melts easily
- D water has color



(a)

Since H atoms are slightly **positive** and the O atoms are slightly **negative**, molecules of water have **polarity**. This makes water a **good** _____

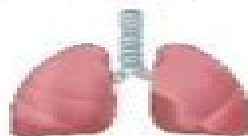
- A solute
- B refrigerant
- C solvent
- D magnet



(c)

The chemical formula for the **oxygen** that we breathe is _____

- A O
- B O_2
- C O_3
- D 2O



(b)

How many **molecules of oxygen** are represented in the chemical formula below?

- A 2
- B 4
- C 6
- D 18



(b)

Chemistry Study Guide Answers Forming Chemical Bonds

David Kirk



Chemistry Study Guide Answers Forming Chemical Bonds:

Reviewing **Chemistry Study Guide Answers Forming Chemical Bonds**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Chemistry Study Guide Answers Forming Chemical Bonds**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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