

NAMES, COMPOUNDS, IONS, + CHARGES TO KNOW

Ag^+ - silver

Zn^{2+} - zinc

* $\text{ic} = \text{more}$, $\text{ous} = \text{less}$ *

ternary ions

Fe^{2+} - ferrous (iron II)

Fe^{3+} ferric (iron III)

* when naming

oxyanion acid H_2

Cu^+ - cuprous (copper I)

Cu^{2+} cupric (copper II)

LIE element - add ide if polyatomic ion, write its name *

phosphite PO_3^{3-}

Sn^{2+} - stannous (tin II)

Sn^{4+} stannic (tin IV)

* acids - $\text{ate} \rightarrow \text{ic}$, $\text{ite} \rightarrow \text{ous}$,

add H^+

Pb^{2+} - plumbous (lead II)

Pb^{4+} plumbic (lead IV)

$\text{ide} \rightarrow \text{hydro} \dots \text{ic}$ *

hydrogen phosphate (olophosphate) HPO_4^{2-}

dihydrogen phosphate H_2PO_4^-

phosphoric acid H_3PO_4

Oxyanions

IV

V

1 more O $\rightarrow$ ate $\rightarrow$ ite $\rightarrow$ 1 less O
* $\text{ate} = \text{more}$, $\text{ite} = \text{less}$ *

has oxygen attached

CO_3^{2-} carbonate

NO_3^- nitrate

SO_4^{2-} sulfate

NO_2^- nitrite

* $\text{ate} + \text{O} = \text{per} \dots \text{ate}$

Miscellaneous ions

$\text{ate} - \text{O} = \text{ite}$

OH^- hydroxide

VI

PO_4^{3-} phosphate

$\text{ite} - \text{O} = \text{hypo} \dots \text{ite}$ *

CH_3COO^- (also $\text{C}_2\text{H}_3\text{O}_2^-$) acetate

Se_6^{2-} selenate

PO_3^{3-} phosphite

CN^- cyanide O_2^{2-} peroxide

SeO_4^{2-} selenate

AsO_4^{3-} arsenate

NH_4^+ ammonium

SeO_3^{2-} selenite

AsO_3^{3-} arsenite

VII

ClO_4^- perchlorate

C	N	O	F
Si	P	S	Cl
Ge	As	Se	Br

- ate ion
has 5 O's
 CO_3^{2-} BrO_3^-
 NO_3^- IO_3^-
 ClO_3^-



ClO_3^- chlorate

ClO_2^- chlorite

OH^- hydroxide

hydrocarbons

BrO_4^- perbromate

C	N	O	F
Si	P	S	Cl
Ge	As	Se	Br

- ate ion
has 4 O's
 PO_4^{3-}
 AsO_4^{3-}
 SO_4^{2-}
 SeO_4^{2-}

carbon + hydrogen (alkanes)

1st part of name = # of carbons

meth-1, eth-2, prop-3...

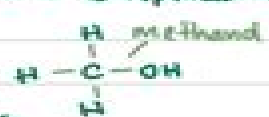
BrO_3^- bromate

BrO_2^- bromite

BrO^- hypobromite

when a H is replaced w/ an OH ends in -ol

IO_4^- periodate



IO_3^- iodate

monatomic ions - first metal (pos) then nonmetal (neg) add -ide ending

OF, LiF - lithium fluoride

IO_2^- iodite

positive		negative (-ide ending)	
Li^+	Be^{2+}	N^{3-}	O^{2-}
Lithium	Beryllium	nitride	oxide
Al^{3+}	Mg^{2+}	P^{3-}	S^{2-}
Aluminum	Magnesium	phosphide	sulfide
			Cl^-
			chloride

BeO_2 - beryllium peroxide

Na_2S - sodium sulfide

IO^- hypoiodite

Chemistry Study Guide Answers Chemical Compounds

Karin Nielsen-Saines



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