

Acids, Bases & pH Review - Honors Chemistry

Directions: For all multiple choice questions, highlight the correct answer(s). Answer short answer questions in the box provided.

Part 1: Properties & definition of Acids & Bases

1. Complete the table below.

	ACIDS	BASES
Taste		
pH range		
Conducting electricity		
Neutralize		
Litmus		
Phenolphthalein		

2. An indicator

- is a type of acid.
- is a type of base.
- has a neutral pH.
- changes color depending on the pH.

3. Litmus and phenolphthalein are indicators. What does that mean?

4. Acids and bases "dissociate" in water to form H^+ (for acids) and OH^- (for bases). What does "dissociate" mean in this context?

5. Identify whether each substance is an acid or a base.

- HBr:
- HCO_2H :
- $Al(OH)_3$:
- $NH_3 + H_2O \rightarrow NH_4^+ + OH^-$; NH_3 is a(n):
- $H_2SO_4 + H_2O \rightarrow H_3O^+ + HSO_4^-$; H_2SO_4 is a(n):
- $HNO_3 \rightarrow H^+ + NO_3^-$; HNO_3 is a(n):

Chemistry Study Guide Acids And Bases

Emilie Sanchez



Chemistry Study Guide Acids And Bases:

Chemistry Study Guide Acids And Bases Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Chemistry Study Guide Acids And Bases**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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