



# CHEMISTRY

## CHEAT SHEET



# STATES OF MATTER

## STATES OF MATTER CHEAT SHEET

Matter		Physical and Chemical Changes	
Matter is anything that takes up space and has volume. It has mass and is made of atoms and molecules.		<p><b>Physical Change:</b> A change in which the substance's physical properties change, but its chemical identity remains the same. Examples: melting, freezing, boiling, condensation, sublimation, deposition.</p> <p><b>Chemical Change:</b> A change in which the substance's chemical identity changes. Examples: burning, rusting, digestion, photosynthesis.</p>	
Phase	Description	Arrangement of Particles	Examples
SOLID	<ul style="list-style-type: none"> <li>Lowest energy state of matter.</li> <li>Particles are closely packed in a regular, repeating pattern.</li> <li>Highly ordered and rigid structure.</li> </ul>		
LIQUID	<ul style="list-style-type: none"> <li>Intermediate state.</li> <li>Particles are close together but can move past each other.</li> <li>Disordered arrangement.</li> </ul>		
GAS	<ul style="list-style-type: none"> <li>Highest energy state of matter.</li> <li>Particles are far apart and move rapidly in all directions.</li> <li>Disordered arrangement.</li> </ul>		
PLASMA	<ul style="list-style-type: none"> <li>Highest energy state of matter.</li> <li>Consists of free electrons and positive ions.</li> </ul>		
<b>Relative Energy Levels:</b> The energy levels of matter are ordered from lowest to highest as follows: Solid, Liquid, Gas, Plasma.		<b>Relative Energy Levels:</b> Solid $\rightarrow$ Liquid $\rightarrow$ Gas $\rightarrow$ Plasma	

# Chemistry States Of Matter Study Guide

**Leo J. Malone, Theodore O. Dolter**



## Chemistry States Of Matter Study Guide:

**Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual** Leo J. Malone, Theodore O. Dolter, 2012-01-03

The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

*Hazmat Chemistry Study Guide (Second Edition)* Jill Meryl Levy, 2011

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14

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