

Chapter 9 Alcohols, Ethers, and Related Compounds

Chapter Review

General facts about ROH, ROR, and epoxides

- All three compounds contain an O atom that is sp^3 hybridized and tetrahedral (9.2).



- All three compounds have polar C-O bonds, but only alcohols have an O-H bond for intermolecular hydrogen bonding (9.4).



- Alcohols and ethers do not contain a good leaving group. Nucleophilic substitution can occur only after the OH (or OR) group is converted to a better leaving group (9.7A).



- Epoxides have a leaving group located in a strained three-membered ring, making them reactive to strong nucleophiles and acids HZ that contain a nucleophilic atom Z (9.16).



A new reaction of carbocations (9.9)

- Less stable carbocations rearrange to more stable carbocations by shift of a hydrogen atom or an alkyl group. Besides rearrangement, carbocations also react with nucleophiles (7.12) and bases (8.6).



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Marc Loudon, Jim Parise, Lisa Bonner



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 - The Rise of Digital Reading Chemistry Solutions Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Solutions Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Solutions Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Solutions Study Guide
 - Personalized Recommendations
 - Chemistry Solutions Study Guide User Reviews and Ratings
 - Chemistry Solutions Study Guide and Bestseller Lists
5. Accessing Chemistry Solutions Study Guide Free and Paid eBooks
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 - Interactive Elements Chemistry Solutions Study Guide
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
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 - Creating a Diverse Reading Collection Chemistry Solutions Study Guide
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 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
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 - Setting Reading Goals Chemistry Solutions Study Guide
 - Carving Out Dedicated Reading Time
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 - Distinguishing Credible Sources
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 - Utilizing eBooks for Skill Development
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