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Chapter 1. Atoms

Student Objectives

- 1.1 A Particulate View of the World: Structure Determines Properties
 - Define atoms, molecules, and the science of chemistry.
 - Represent a simple molecule, water, using spheres as atoms.
- 1.2 Classifying Matter: A Particulate View
 - Define matter and distinguish between the three main states of matter: solid, liquid, and gas.
 - Define and understand the difference between crystalline and amorphous solids.
 - Define mixture, pure substance, element, compound, heterogeneous, and homogeneous.
 - Differentiate between mixtures and pure substances; elements and compounds; and heterogeneous and homogeneous mixtures.
 - Use the scheme on page 000 to classify matter.
- 1.3 The Scientific Approach to Knowledge
 - Define and distinguish between a hypothesis, a scientific law, and a theory.
 - Understand the role of experiments in testing hypotheses.
 - State and understand the law of mass conservation as an example of scientific law.
 - Understand that scientific theories are built from strong experimental evidence and that the term "theory" in science is used much differently than in pop culture.
 - Understand the importance of reporting correct units with measurements.

Chemistry Solution Manual Tro

Nivaldo J. Tro, Travis Fridgen, Lawton Shaw

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