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### Chapter 3 – Scientific Measurement Study Guide

- Qualitative vs. Quantitative measurements
- Scientific notation → decimal number
  - Ex.)  $3.45 \times 10^{-5}$
  - Ex.)  $6.02 \times 10^8$
- Decimal number → scientific notation
  - Ex.) 450,000,000
  - Ex.) .0000000045
- Multiply and divide numbers in scientific notation
  - What are the rules? What do you do with the coefficient? What do you do with the exponent?
  - Ex.)  $(5 \times 10^{-2}) \times (11 \times 10^6)$
  - Ex.)  $(8 \times 10^6) / (4 \times 10^5)$
- Add and subtract numbers in scientific notation
  - What are the rules? What do you do with the coefficient? What do you do with the exponent?
  - Ex.)  $(4 \times 10^5) + (3 \times 10^5)$
  - Ex.)  $(2 \times 10^2) - (4 \times 10^2)$
- % error (know how to use the formula)
  - Ex.) If the observed value for a density is 0.80 g/ml and the accepted value is 0.70 g/ml, what is the percent error?

# Chemistry Scientific Measurement Study Guide Answers

**Sebastian Brünink**



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