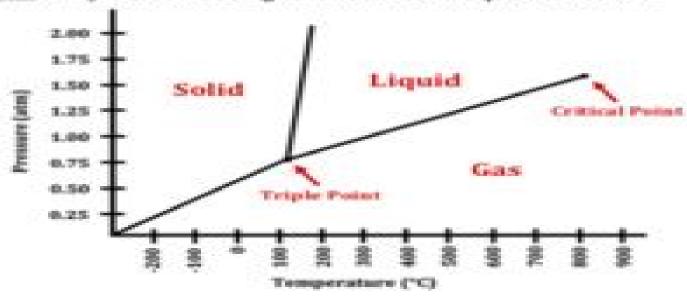
Phase Diagram

Instructions: Analyze the Phase Diagram and answer the questions below:



- 1) In the diagram above, label the areas that represent solid. Books, and gas phases of matter.
- 2) At the <u>triple point</u>, all three states of matter exist together. Label the <u>triple point</u> above.
- 1) At approximately what pressure does the triple point occur? 0.75 atm
- 45 At approximately what temperature does the triple point occur? 100°C
- 5) At the <u>critical point</u> liquid and gas become indistinguishable. Label the <u>critical point</u> above.
- 6) At approximately what pressure does the critical point occur? 1.60 atm
- 7). At approximately what temperature does the critical point occur? 800°C
- II). At 300°C and 1.25 atm of pressure, what phase would the substance be in? Liquid
- 9) At 600°C and 0.50 atm of pressure, what phase would the substance be in? Gas
- 10) At -100°C and 0.50 atm of pressure, what phase would the substance be in? Solid

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