

Periodic Table Trend Activities:



Purpose: The Periodic Table is a useful tool that helps us predict properties of various elements. In this activity, we will look at 3 properties:

1. **Atomic radius:** The distance from the atomic nucleus to the outmost occupied orbital in a atom
2. **Electronegativity:** The ability of an atom to attract a bonded pair of electrons from another atom
3. **Ionization Energy:** The energy needed to remove an electron from a neutral gaseous atom
4. **Electron Affinity:** The energy released when an electron is added to a neutral gaseous atom

Procedure: Click on the [Cool Periodic Table link](http://www.lynchburg.net/hhs/chemistry/trends/) (click on the element symbol). Record the atomic radius, electronegativity, and first ionization energy for each element in the table below:

<http://www.lynchburg.net/hhs/chemistry/trends/>

Data:

Element	Atomic Radius	Electronegativity	1 st Ionization Energy
Li			
Be			
B			
Na			
Mg			
Al			
C			
Si			
N			
P			

Discussion:

1. What appears to be the trend in atomic radius as you move from left to right in a row?
2. What appears to be the trend in atomic radius as you move down a column?

Chemistry Periodic Trends Lab Activity

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Table of Contents Chemistry Periodic Trends Lab Activity

1. Understanding the eBook Chemistry Periodic Trends Lab Activity
 - The Rise of Digital Reading Chemistry Periodic Trends Lab Activity
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Periodic Trends Lab Activity
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Periodic Trends Lab Activity
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Periodic Trends Lab Activity
 - Personalized Recommendations
 - Chemistry Periodic Trends Lab Activity User Reviews and Ratings
 - Chemistry Periodic Trends Lab Activity and Bestseller Lists
5. Accessing Chemistry Periodic Trends Lab Activity Free and Paid eBooks
 - Chemistry Periodic Trends Lab Activity Public Domain eBooks
 - Chemistry Periodic Trends Lab Activity eBook Subscription Services

- Chemistry Periodic Trends Lab Activity Budget-Friendly Options
- 6. Navigating Chemistry Periodic Trends Lab Activity eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Periodic Trends Lab Activity Compatibility with Devices
 - Chemistry Periodic Trends Lab Activity Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Periodic Trends Lab Activity
 - Highlighting and Note-Taking Chemistry Periodic Trends Lab Activity
 - Interactive Elements Chemistry Periodic Trends Lab Activity
- 8. Staying Engaged with Chemistry Periodic Trends Lab Activity
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Periodic Trends Lab Activity
- 9. Balancing eBooks and Physical Books Chemistry Periodic Trends Lab Activity
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Periodic Trends Lab Activity
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Periodic Trends Lab Activity
 - Setting Reading Goals Chemistry Periodic Trends Lab Activity
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Periodic Trends Lab Activity
 - Fact-Checking eBook Content of Chemistry Periodic Trends Lab Activity
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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