

Chapter 2 Chemistry of Life

- Matter anything that takes up space
- An Element is one of the basic building blocks of matter and cannot be broken down by chemical means.
- It is even more surprising that that over 90% of the human body is composed of just four elements.
 - Carbon
 - Nitrogen
 - Oxygen
 - Hydrogen
- Atoms are the simplest unit of element that still retains the chemical and physical properties of an element.
 - There are the best-known subatomic particles are positively charged protons
 - Uncharged: Neutrons
 - Negatively charged: Electrons
 - Protons and Neutrons are located within the nucleus of an atom and electrons move about the nucleus
 - The circle around the nucleus of an atom represents an electron shell, which represents the average location of electrons
- The atomic number represents the number of protons in the nucleus of an element
- Mass number of an atom represents the sum of protons and neutrons in the nucleus
- The atomic mass (the number below the atomic symbol on the periodic table) is the average of the AMU for all the isotopes of that atom. To determine the number of neutrons, subtract the number of protons from the atomic mass and take the closest whole number.
- Isotopes of an atom have the same number of protons but different numbers
- Atom with more than one shell are most stable when the outer shell is also called shell contains 8 electrons.
- Ions are particles that either carry either positive or negative charge.
- The attraction between oppositely charged sodium ions and chloride ions form an ionic bond.
- Atoms share in covalent bonds
- A hydrogen bond is the attraction of a slightly positive covalently bonded hydrogen to a slightly negative atom in the vicinity
- Water has a high heat capacity
- Water has a high heat of evaporation
- Water molecules are cohesive and adhesive
- Cohesion refers to the ability of water molecules to cling to each other
- Adhesion refers to the ability of water molecule to cling to a surface
- Lemon juice, Vinegar, Tomatoes, and coffee are all acidic solutions
- Acids are substances that dissociates in water releasing hydrogen ions. The acidity of a substance depends on how fully it dissociates in water

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Table of Contents Chemistry Of Life Biology Study Guide Answers

1. Understanding the eBook Chemistry Of Life Biology Study Guide Answers
 - The Rise of Digital Reading Chemistry Of Life Biology Study Guide Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Of Life Biology Study Guide Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Of Life Biology Study Guide Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Of Life Biology Study Guide Answers
 - Personalized Recommendations
 - Chemistry Of Life Biology Study Guide Answers User Reviews and Ratings
 - Chemistry Of Life Biology Study Guide Answers and Bestseller Lists
5. Accessing Chemistry Of Life Biology Study Guide Answers Free and Paid eBooks

- Chemistry Of Life Biology Study Guide Answers Public Domain eBooks
 - Chemistry Of Life Biology Study Guide Answers eBook Subscription Services
 - Chemistry Of Life Biology Study Guide Answers Budget-Friendly Options
6. Navigating Chemistry Of Life Biology Study Guide Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Of Life Biology Study Guide Answers Compatibility with Devices
 - Chemistry Of Life Biology Study Guide Answers Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Of Life Biology Study Guide Answers
 - Highlighting and Note-Taking Chemistry Of Life Biology Study Guide Answers
 - Interactive Elements Chemistry Of Life Biology Study Guide Answers
 8. Staying Engaged with Chemistry Of Life Biology Study Guide Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Of Life Biology Study Guide Answers
 9. Balancing eBooks and Physical Books Chemistry Of Life Biology Study Guide Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Of Life Biology Study Guide Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Chemistry Of Life Biology Study Guide Answers
 - Setting Reading Goals Chemistry Of Life Biology Study Guide Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Chemistry Of Life Biology Study Guide Answers
 - Fact-Checking eBook Content of Chemistry Of Life Biology Study Guide Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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