natural transmutation - Uranium spontaneously decays.

$$^{238}_{92}U \rightarrow ^{234}_{90}Th + ^{4}_{2}He$$

<u>artificial transmutation</u> - bombardment of a stable isotope to force it to decay.

$$^{14}_{7}N + ^{4}_{2}He \rightarrow ^{1}_{1}p + ^{17}_{6}D$$

When the bullets are + charged, they are repelled by the nucleus they are bombarding. To overcome the repulsions, they must be outstand to very high speeds by Posticia

nuclear fission - Heavy nuclei are bombarded with neutrons and split. 1 1 ^{235}II $> ^{142}D$ = 1 ^{91}V = 1 = 2 1 = 1

$$_{0}^{1}n + _{92}^{235}U \rightarrow _{56}^{142}Ba + _{36}^{91}Kr + 3(_{0}^{1}n) + E$$

Mass of particles produced is slightly <u>more</u> than the mass of the reactants.

This mass is converted into <u>energy</u>. (E = m c²)

required for a Chain sealtion

nuclear reactors: control fission <u>Chain</u> reactions to produce energy dangers:

nuclear fusion - combination of 2 nuclei into 1 with release of

$${}_{1}^{2}H + {}_{1}^{3}H \rightarrow {}_{2}^{4}He + {}_{0}^{1}n + E$$

Mass of particles produced is much MOLE than the mass of the ractions.

This Mass is converted into energy. (E = Mac)

On back, list advantages of and problems with using fusion as an energy source.

CHEMISTRY: A Study of Matter

The issues with using fusion as energy generally just come from concern for the public and their safety due to storage of nuclear material for a long time causing concerns well as radioactivity being a big issue.

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- 1. Understanding the eBook Chemistry Note Taking Guide 1502
 - The Rise of Digital Reading Chemistry Note Taking Guide 1502
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Chemistry Note Taking Guide 1502
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Note Taking Guide 1502
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chemistry Note Taking Guide 1502
 - Personalized Recommendations
 - Chemistry Note Taking Guide 1502 User Reviews and Ratings
 - Chemistry Note Taking Guide 1502 and Bestseller Lists

- 5. Accessing Chemistry Note Taking Guide 1502 Free and Paid eBooks
 - Chemistry Note Taking Guide 1502 Public Domain eBooks
 - Chemistry Note Taking Guide 1502 eBook Subscription Services
 - Chemistry Note Taking Guide 1502 Budget-Friendly Options
- 6. Navigating Chemistry Note Taking Guide 1502 eBook Formats
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 - Chemistry Note Taking Guide 1502 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
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 - Highlighting and Note-Taking Chemistry Note Taking Guide 1502
 - Interactive Elements Chemistry Note Taking Guide 1502
- 8. Staying Engaged with Chemistry Note Taking Guide 1502
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - \circ Following Authors and Publishers Chemistry Note Taking Guide 1502
- 9. Balancing eBooks and Physical Books Chemistry Note Taking Guide 1502
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Note Taking Guide 1502
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Note Taking Guide 1502
 - Setting Reading Goals Chemistry Note Taking Guide 1502
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Note Taking Guide 1502
 - Fact-Checking eBook Content of Chemistry Note Taking Guide 1502
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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