

CHAPTER **11** STUDY GUIDE

Stoichiometry

Section 11.1 What is stoichiometry?

In your textbook, read about stoichiometry and the balanced equation.

For each statement below, write *true* or *false*.

- _____ 1. The study of the quantitative relationships between the amounts of reactants used and the amounts of products formed by a chemical reaction is called stoichiometry.
- _____ 2. Stoichiometry is based on the law of conservation of mass.
- _____ 3. In any chemical reaction, the mass of the products is less than the mass of the reactants.
- _____ 4. The coefficients in a chemical equation represent not only the number of individual particles but also the number of moles of particles.
- _____ 5. The mass of each reactant and product is related to its coefficient in the balanced chemical equation for the reaction by its molar mass.

Complete the table below, using information represented in the chemical equation for the combustion of methanol, an alcohol.



Substance	Molar Mass (g/mol)	Number of Molecules	Number of Moles (mol)	Mass (g)
6. Methanol	32.05			
7. Oxygen gas	32.00			
8. Carbon dioxide	44.01			
9. Water	18.02			

10. What are the reactants? _____
11. What are the products? _____
12. What is the total mass of the reactants? _____
13. What is the total mass of the products? _____
14. How do the total masses of the reactants and products compare? _____

Chemistry Matter And Change Chapter 11 Study Guide

Karin Nielsen-Saines



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Whispering the Strategies of Language: An Emotional Journey through **Chemistry Matter And Change Chapter 11 Study Guide**

In a digitally-driven world wherever screens reign great and immediate interaction drowns out the subtleties of language, the profound strategies and mental nuances hidden within words frequently move unheard. Yet, set within the pages of **Chemistry Matter And Change Chapter 11 Study Guide** a interesting fictional prize sporting with fresh feelings, lies a fantastic journey waiting to be undertaken. Penned by an experienced wordsmith, that enchanting opus invites viewers on an introspective journey, delicately unraveling the veiled truths and profound influence resonating within the cloth of each and every word. Within the psychological depths of this emotional evaluation, we can embark upon a heartfelt exploration of the book is key subjects, dissect their charming publishing fashion, and succumb to the powerful resonance it evokes serious within the recesses of readers hearts.

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