

SYLLABUS

Time Allowed: 3 Hours

Max. Marks: 30

Evaluation Scheme for Examination	Marks
• Volumetric Analysis	08
• Salt Analysis	08
• Content Based Experiment	06
• Project work	04
• Class Record and Viva	04
Total	30

PRACTICAL SYLLABUS

Total Periods: 60

Micro-chemical methods are available for several of the practical experiments. Wherever possible, such techniques should be used.

A. Surface Chemistry

- Preparation of one lyophilic and one lyophobic sol.
Lyophilic sol-starch, egg albumin and gum
Lyophobic sol-aluminum hydroxide, ferric hydroxide and arsenous sulphide
- Dialysis of sol-prepared in (a) above.
- Study of the role of emulsifying agents in stabilizing the emulsion of different oils.

B. Chemical Kinetics

- Effect of concentration and temperature on the rate of reaction between Sodium Thiosulphate and Hydrochloric acid.
- Study of reaction rates of any one of the following:
 - Reaction of Iodide ion with Hydrogen Peroxide at room temperature using different concentration of Iodide ions.
 - Reaction between Potassium Iodate (KIO_3), and Sodium Sulphite (Na_2SO_3) using starch solution as indicator (clock reaction).

C. Thermochemistry

- Any one of the following experiments:
- Enthalpy of dissolution of Copper Sulphate or Potassium Nitrate.
 - Enthalpy of neutralization of a strong acid (HCl) and a strong base (NaOH).
 - Determination of enthalpy change during interaction (Hydrogen bond formation) between Acetone and Chloroform.

D. Electrochemistry

- Variation of cell potential in $\text{Zn}/\text{Zn}^{2+} \parallel \text{Cu}^{2+}/\text{Cu}$ with change in concentration of electrolytes (CuSO_4 or ZnSO_4) at room temperature.

E. Chromatography

- Separation of pigments from extracts of leaves and flowers by paper chromatography and determination of R_f value
- Separation of constituents present in an inorganic mixture containing two cations only (constituents having large difference in R_f values to be provided).

Chemistry Lab Manual Class 12 State Board

Peipei Pang



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