

## CHEMISTRY MID-TERM STUDY GUIDE

For questions 1-6, determine the number of significant figures in each measurement.

- \_\_\_\_\_1) 4658 g                      \_\_\_\_\_3) 0.0020 km                      \_\_\_\_\_5) 200 m  
\_\_\_\_\_2) 360.0 cm                      \_\_\_\_\_4)  $7.02 \times 10^5$  mm                      \_\_\_\_\_6) 8050. S

For questions 7-10, determine the number of significant figures in the calculated answers without solving the problems.

- \_\_\_\_\_7)  $4.033 \times 8.99 =$                       \_\_\_\_\_9)  $34.6 + 22.22 =$   
\_\_\_\_\_8)  $0.330 / 5.6 =$                       \_\_\_\_\_10)  $22899 - 2 =$   
\_\_\_\_\_11) In the measurement 305.67g, which digit is estimated?

a. The 3                      b. The 0                      c. The 6                      d. The 7

\_\_\_\_\_12) A piece of paper measures 220.5 cm by 76.3 cm. What is the area of this piece of paper expressed to the correct number of sig. figs.?

a.  $16824.15 \text{ cm}^2$                       b.  $16824 \text{ cm}^2$                       c.  $16820 \text{ cm}^2$                       d.  $16800 \text{ cm}^2$

\_\_\_\_\_13) A sample of metal is found to have a density of 8.44 g/mL and a mass of 10.00g. What is the volume of this metal?

a. 1.18 mL                      b. 1.180 mL                      c. 0.844 mL                      d. 0.84 mL

\_\_\_\_\_14) Calculate the density of a gas that has a volume of  $1.033 \times 10^{-2}$  L and a mass of  $4.22 \times 10^{-4}$  g.

a.  $4.09 \times 10^2 \text{ g/L}$                       b.  $4.09 \times 10^{-2} \text{ g/L}$                       c.  $2.45 \times 10^2 \text{ L/g}$

For questions 15 - 18 choose "A" if the statement refers to an ionic compound and "B" if the statement refers to a covalent compound.

- \_\_\_\_\_15) Exist as neutral molecules.  
\_\_\_\_\_16) Made from a metallic and a nonmetallic element.  
\_\_\_\_\_17) Made from 2 or more nonmetallic elements.  
\_\_\_\_\_18) A crystal lattice structure with a very high melting point.  
\_\_\_\_\_19) How many dots does nitrogen have in its electron dot diagram?  
a. 2                      b. 4                      c. 5                      d. 7

# Chemistry Honors Midterm Study Guide

**Steven S. Zumdahl, Susan A.  
Zumdahl, Donald J. DeCoste**



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