

CHEMISTRY FINAL EXAM REVIEW GUIDE 2012 Scott

DISCLAIMER: THIS IS A SAMPLE OF QUESTIONS/PROBLEMS FROM THIS YEAR. PLEASE BE SURE TO REVIEW YOUR NOTES, LABS, COJ'S, OBJECTIVE SHEETS, ETC. DO NOT RELY ONLY ON THIS REVIEW GUIDE.

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1) Which of the following observations is quantitative?
a. The liquid turns blue litmus paper red.
b. The liquid boils at 100°C.
c. The liquid tastes bitter.
d. The liquid is cloudy.
- _____ 2) The symbols for units of length in order from largest to smallest are
a. m, cm, mm, km. c. km, mm, cm, m.
b. mm, m, cm, km. d. km, m, cm, mm.
- _____ 3) A volume of 1 milliliter is equivalent to
a. 1 cubic centimeter. c. 1 liter.
b. 1 gram. d. 10^{-3} cubic decimeters.
- _____ 4) Which pair of quantities determines the density of a material?
a. mass and weight c. volume and concentration
b. volume and weight d. volume and mass
- _____ 5) A sample of gold has a mass of 96.5 g and a volume of 5.00 cm³. The density of gold is
a. 0.0518 g/cm³. c. 101.5 g/cm³.
b. 19.3 g/cm³. d. 483 g/cm³.
- _____ 6) The number of grams equal to 0.5 kg is
a. 0.0005. c. 500.
b. 0.005. d. 5000.
- _____ 7) If 1 inch equals 2.54 cm, how many centimeters equal 1 yard?
a. 0.0706 cm c. 30.5 cm
b. 14.2 cm d. 91.4 cm
- _____ 8) A chemist obtained the values 5.2246 g, 5.2353 g, and 5.2501 g for the mass of a sample. Without knowing the true mass of the sample, it can be said that these values have
a. good precision.
b. good accuracy.
c. poor precision.
d. poor accuracy.
- _____ 9) The measurement 0.035550 g rounded off to two significant figures would be
a. 0.03 g. c. 0.036 g.
b. 0.35 g. d. 3.5×10^2 g.
- _____ 10) When 64.4 is divided by 2.00, the correct number of significant figures in the result is

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Table of Contents Chemistry Final Exam Review Guide

1. Understanding the eBook Chemistry Final Exam Review Guide
 - The Rise of Digital Reading Chemistry Final Exam Review Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Final Exam Review Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Final Exam Review Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Final Exam Review Guide
 - Personalized Recommendations
 - Chemistry Final Exam Review Guide User Reviews and Ratings
 - Chemistry Final Exam Review Guide and Bestseller Lists
5. Accessing Chemistry Final Exam Review Guide Free and Paid eBooks
 - Chemistry Final Exam Review Guide Public Domain eBooks
 - Chemistry Final Exam Review Guide eBook Subscription Services
 - Chemistry Final Exam Review Guide Budget-Friendly Options
6. Navigating Chemistry Final Exam Review Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Final Exam Review Guide Compatibility with Devices
 - Chemistry Final Exam Review Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Final Exam Review Guide
 - Highlighting and Note-Taking Chemistry Final Exam Review Guide
 - Interactive Elements Chemistry Final Exam Review Guide
8. Staying Engaged with Chemistry Final Exam Review Guide

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Chemistry Final Exam Review Guide
- 9. Balancing eBooks and Physical Books Chemistry Final Exam Review Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Final Exam Review Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Final Exam Review Guide
 - Setting Reading Goals Chemistry Final Exam Review Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Final Exam Review Guide
 - Fact-Checking eBook Content of Chemistry Final Exam Review Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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