

Chemistry Cookie Project –Chocolate Chip

- In this lab you will be converting a recipe from moles to standard cooking measurements and then using that recipe to bake some cookies!
- You will need the following tables in order to convert your recipe:

Molecular Formula Table

Use the following molecular formulas to calculate grams of ingredients.
NOTE: Most of these substances have extremely complex molecular formulas.
 I have greatly simplified your project by listing a representative formula only.

Alum	$\text{C}_6\text{H}_{12}\text{O}_6$
Baking Soda	NaHCO_3
Baking Powder	NaHCO_3
Brown Sugar	$\text{C}_{12}\text{H}_{22}\text{O}_{11}$
Butter	$\text{C}_8\text{H}_{16}\text{O}_8$
White (Granu) Sugar	$\text{C}_{12}\text{H}_{22}\text{O}_{11}$
Chocolate	$\text{C}_8\text{H}_8\text{O}_4$
Cinnamon	$\text{C}_9\text{H}_{10}\text{O}$
Chocis	$\text{C}_{10}\text{H}_{12}\text{O}_5$
Cream of tartar	$\text{KHC}_4\text{H}_7\text{O}_6$

Flour	$\text{C}_6\text{H}_{12}\text{O}_6$
Lemon Juice (Citrus)	$\text{C}_6\text{H}_8\text{O}_7$
Margarine	$\text{C}_{18}\text{H}_{34}\text{O}_2$
Milk	85% H_2O 15% $\text{C}_8\text{H}_{16}\text{O}_8$
Molasses	$\text{C}_{12}\text{H}_{22}\text{O}_{11}$
Pepper	$\text{C}_{11}\text{H}_{15}\text{O}_5\text{N}$
Salt	NaCl
Vanilla	$\text{C}_8\text{H}_8\text{O}_3$
Eggs	$\text{C}_6\text{H}_{12}\text{O}_6\text{N}_2$

Unit Conversions

Use the following conversions to get from grams to a standard cooking unit of measure.

1 teaspoon of baking soda = 2.84g
 1 teaspoon of vanilla extract = 4.73g
 1 large egg = 50g
 1 cup flour = 141.95 g
 1 teaspoon salt = 4.16 g
 1 cup butter = 236.59 g
 1 cup sugar = 198.73 g
 1 cup brown sugar = 141.46 g
 1 ounce chocolate chips = 28.35 g
 1 tablespoon lemon juice = 14.20 g
 1 teaspoon baking powder = 2.84 g

1 teaspoon cream of tartar = 2.84 g
 1 teaspoon cinnamon = 2.84 g

Chemistry Cookie Project Chocolate Chip Answers

Robert R. Redfield



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