

# CHAPTER 14 STUDY GUIDE

## Mixtures and Solutions

### Section 14.1 Heterogeneous and Homogeneous Mixtures

*At your notebook, read about suspensions and colloids.*

*For each statement below, write true or false.*

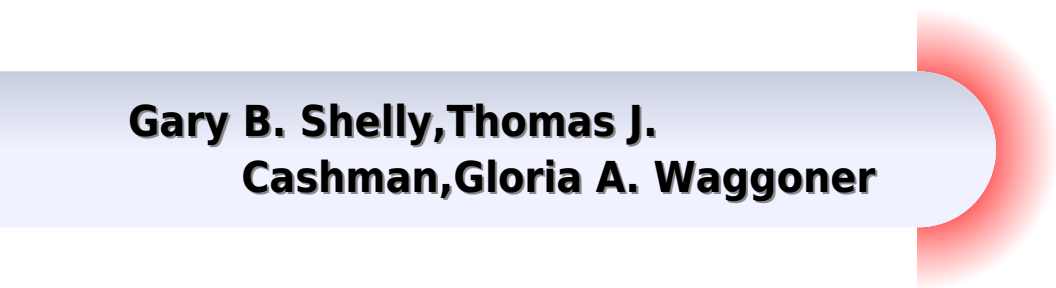
- \_\_\_\_\_ 1. A solution is a mixture consisting particles that settle out of the mixture if left undisturbed.
- \_\_\_\_\_ 2. The most abundant substance in a colloid is the dispersion medium.
- \_\_\_\_\_ 3. A colloid can be separated by filtration.
- \_\_\_\_\_ 4. A solid emulsion consists of a liquid dispersed in a solid.
- \_\_\_\_\_ 5. Whipped cream is an example of a foam.
- \_\_\_\_\_ 6. In an aerosol, the dispersing medium is a liquid.
- \_\_\_\_\_ 7. Brownian motion results from the collisions of particles of the dispersion medium with the dispersed particles.
- \_\_\_\_\_ 8. Dispersed particles in a colloid do not tend to settle out because they have polar or charged atomic groups on their surfaces.
- \_\_\_\_\_ 9. Adding an electrolyte into a colloid stabilizes the colloid.
- \_\_\_\_\_ 10. Colloids demonstrate the Tyndall effect.

The table below lists the characteristics of particles in colloids, solutions, and suspensions. Place a check in the column of each mixture where particles have a particular characteristic.

| Characteristics of Particles             | Colloid | Solution | Suspension |
|------------------------------------------|---------|----------|------------|
| 01. Less than 1 nm in diameter           |         |          |            |
| 02. Between 1 nm and 1000 nm in diameter |         |          |            |
| 03. More than 1000 nm in diameter        |         |          |            |
| 04. Settle out if undisturbed            |         |          |            |
| 05. Pass through standard filter paper   |         |          |            |
| 06. Largest visible particles            |         |          |            |
| 07. Scatter light                        |         |          |            |

# Chemistry Chapter 14 Study Guide For Content Mastery Answers

**Gary B. Shelly, Thomas J.  
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## **Table of Contents Chemistry Chapter 14 Study Guide For Content Mastery Answers**

1. Understanding the eBook Chemistry Chapter 14 Study Guide For Content Mastery Answers
  - The Rise of Digital Reading Chemistry Chapter 14 Study Guide For Content Mastery Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Chapter 14 Study Guide For Content Mastery Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Chapter 14 Study Guide For Content Mastery Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Chapter 14 Study Guide For Content Mastery Answers
  - Personalized Recommendations
  - Chemistry Chapter 14 Study Guide For Content Mastery Answers User Reviews and Ratings

- Chemistry Chapter 14 Study Guide For Content Mastery Answers and Bestseller Lists
- 5. Accessing Chemistry Chapter 14 Study Guide For Content Mastery Answers Free and Paid eBooks
  - Chemistry Chapter 14 Study Guide For Content Mastery Answers Public Domain eBooks
  - Chemistry Chapter 14 Study Guide For Content Mastery Answers eBook Subscription Services
  - Chemistry Chapter 14 Study Guide For Content Mastery Answers Budget-Friendly Options
- 6. Navigating Chemistry Chapter 14 Study Guide For Content Mastery Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Chapter 14 Study Guide For Content Mastery Answers Compatibility with Devices
  - Chemistry Chapter 14 Study Guide For Content Mastery Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Chapter 14 Study Guide For Content Mastery Answers
  - Highlighting and Note-Taking Chemistry Chapter 14 Study Guide For Content Mastery Answers
  - Interactive Elements Chemistry Chapter 14 Study Guide For Content Mastery Answers
- 8. Staying Engaged with Chemistry Chapter 14 Study Guide For Content Mastery Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Chapter 14 Study Guide For Content Mastery Answers
- 9. Balancing eBooks and Physical Books Chemistry Chapter 14 Study Guide For Content Mastery Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Chapter 14 Study Guide For Content Mastery Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Chapter 14 Study Guide For Content Mastery Answers
  - Setting Reading Goals Chemistry Chapter 14 Study Guide For Content Mastery Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Chapter 14 Study Guide For Content Mastery Answers
  - Fact-Checking eBook Content of Chemistry Chapter 14 Study Guide For Content Mastery Answers
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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