

Name _____ Date _____

Chapter 12 – Stoichiometry Study Guide

The following example problems exhibit the types of calculations you will be expected to perform on tomorrow's test.

- Describe a reaction in terms of particles.
Ex.) Describe the following reaction in terms of molecules



* Note: sometimes you will need to include other descriptors such as atoms and formula units.*

- Describe a reaction in terms of moles.
Ex.) Describe the following reaction in terms of moles



- How do you determine the mass of the products for a balanced reaction?
Ex.) What is the mass of the product formed in the following equation?



- How do you determine the volume of the reactants or products of a balanced reaction?
Ex.) What is the volume of the products and the reactants of the following equation?



- How do you show how a balanced reaction follows the Law of Conservation of Mass by showing that the mass on both sides of a reaction are equal?

Ex.) Show how the mass of the reactants equals the mass of the products for the following equation.

Chemistry Chapter 12 Stoichiometry Study Guide

Arshad Iqbal



Chemistry Chapter 12 Stoichiometry Study Guide:

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