

1. 6. Which condition is necessary for most gases to behave nearly ideally?
- a. high pressure
  - b. high temperature
  - c. low compressibility
  - d. low expansion
7. Which of the following is *not* a physical property of gases?
- a. absence of definite volume
  - b. large density
  - c. high compressibility
  - d. fluidity
8. Rates of effusion of different gases are proportional to their
- a. polarity.
  - b. particle charge.
  - c. particle velocities.
  - d. compressibility.
9. Which states of matter are fluid?
- a. gases and liquids
  - b. liquids and solids
  - c. gases only
  - d. liquids only
10. Which gas deviates the most from ideal behavior?
- a.  $\text{CH}_4$
  - b.  $\text{H}_2\text{O}$
  - c.  $\text{Cl}_2$
  - d. He

# Chemistry Ch 10 Section 1 Review Answers

**Thomas Evangelist**



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