

Energy and Chemical Change

Section 16.1 Energy

In your textbook, read about the nature of energy.

In the space at the left, write *true* if the statement is true; if the statement is false, change the italicized word or phrase to make it true.

- | | |
|-------|---|
| _____ | 1. Energy is the ability to do work or produce heat. |
| _____ | 2. The law of conservation of energy states that energy <i>can be</i> created and destroyed. |
| _____ | 3. Chemical potential energy is energy stored in a substance because of its <i>composition</i> . |
| _____ | 4. <i>Heat</i> is a form of energy that flows from a warmer object to a cooler object. |
| _____ | 5. A calorie is the amount of energy required to raise the temperature of <i>one gram</i> of pure water by one degree Celsius. |
| _____ | 6. A <i>calorie</i> is the SI unit of heat and energy. |
| _____ | 7. The <i>specific heat</i> of a substance is the amount of heat required to raise the temperature of one gram of that substance by one degree Celsius. |
| _____ | 8. <i>Kinetic energy</i> is energy of motion. |
| _____ | 9. Chemicals participating in a chemical reaction contain <i>only potential energy</i> . |
| _____ | 10. One nutritional Calorie is equal to <i>100 calories</i> . |
| _____ | 11. One calorie equals <i>4.184 joules</i> . |
| _____ | 12. When a fuel is burned, some of its <i>chemical potential energy</i> is lost as heat. |
| _____ | 13. To convert kilojoules to joules, <i>divide</i> the number of kilojoules by 1000 joules/1 kilojoule. |

Answer the following question. Show all your work.

14. If the temperature of a 500.0-g sample of liquid water is raised 2.00°C, how much heat is absorbed by the water? The specific heat of liquid water is 4.184 J/(g·°C).

Chemistry And 16 Study Guide Answers

**John Krenos, Joseph Potenza, Loretta
Jones, Lynn Koplitz, Thomas Spence**



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below. Match the correct ... Mark Schemes Exploring Science edition. © Pearson Education Limited 2008. 187. 8. F. Quick Quiz 1 ... Matching End of Unit Test marks to NC levels. Level Marks available. Year 8 Unit 8F End of Unit Quick Quiz | 52 plays Year 8 Unit 8F End of Unit Quick Quiz quiz for 8th grade students. Find other quizzes for Chemistry and more on Quizizz for free! Get Exploring Science 8f End Of Unit Test Complete Exploring Science 8f End Of Unit Test online with US Legal Forms. Easily fill out PDF blank, edit, and sign them. Save or instantly send your ready ... year-8-assessment-support-sample-unit-8hb.pdf End of Unit Test Mark Scheme Standard (S). Question Part Level Answer. Mark scheme. 1. 3. Any two from: colour, textures, hardness/ crumbliness, porous, layers ... End of Unit Test 1 Here are the names of some substances. sulphur copper oxygen iron water magnesium mercury. Which substance: a is a gas at room temperature? Revision 8F Periodic Table (Exploring Science) Nov 25, 2019 — This revision mat covers Unit 8F of Exploring Science: Periodic Table. It includes all of the topics in the book. The revision mat is great ...