

NAME: _____

Chem 305 Practice for Final

- Perform the following operations and round your answers to the correct number of significant digits.
 - $2.354 - 2.154$
 - $26.0 \text{ cm}^2 + 10.1 \text{ cm}$
- Convert the following and express your answers in scientific notation.
 - 378 Mm to m
 - 35000 mg to kg
- Water contains 11.2% hydrogen and 88.8% oxygen. What mass of water contains 50.0 g of oxygen? BP Answer: 50-60 g H₂O
- What is the mass of a rectangular object measuring 0.025 m by 2.5 cm by 5.6 cm and has a density of 3.86 g/cm³? BP Answer: 200 g
- State the number of neutrons in each of the following.
 - iodine-127
 - ${}_{36}^{78}\text{Kr}$
- Silver occurs naturally as ¹⁰⁷Ag and ¹⁰⁹Ag. Given the mass and abundance of silver-109 (108.91 amu and 48.16%), what is the isotopic mass of silver-107? BP Answer: 107 amu
- State which of the following are physical or chemical properties or changes.
 - sugar dissolved in water
 - Manganese bubbles in acid
 - Silver is malleable
- What is the total number of orbitals in the n=2 energy level? Indicate the type of orbitals and how many of each type of orbital. Indicate the total number of electrons that can reside in n=2.
- Write the electron configuration for Mn²⁺
- Refer to the periodic table to answer the following:
 - The name of the alkaline earth metal in the third energy level
 - The name and symbol of the element in group VA that is a gas at room temp and pressure

Chemistry 305 Final Exam

Michael Rajnik



Chemistry 305 Final Exam:

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