

LABORATORY DATA SHEET

Experiment 2

Name _____

Density

Date & Section _____

6. Calculate the % error in the copper wire density 6.0 %

7. Which calculated density of copper (cylinder or wire) do you think is more accurate? Why?
 wire, Percent error is smaller than the copper.

Density of Water:

	Trial 1	Trial 2	Trial 3
Mass of empty beaker (g)	0.9	28.9	28.7
Initial volume reading (ml)	0.2	0.3	2.5
Final volume reading (ml)	21	38mL	40.04
Calculated volume (ml)	3.2	37.7	38.9
Total mass of beaker & water (g)	4.2	66.4	20.4
Mass of water only (g)	23.2	34.3	31.0
Temperature of water (°C)	24C	21C	20.3
Calculated density of water (g/ml)	1.6	0.99	1.00

*** For Trial 2 and 3 get the data from a different group.*

1. Calculate the average temperature of the water 22 °C

2. Calculate the average density of the water 1.20 g/ml

3. According to Appendix B, what is the accepted density of water at your average temperature? 20.1 g/ml

4. Calculate the % error of your measurement by comparing the accepted density with your average density. 20.27 %

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