# Chemistry 104 Reference Sheet Naming Organic Molecules

Alkenes: \_\_\_ene

How these are named:

- 1) The longest C-chain with the double bond is the main chain.
- 2) Number so that the double bond gets the lowest possible number.
- 3) If the structure is a cis-trans (geometric) isomer, precede the name with cis or trans.

Example:

Cycloalkenes: cyclo\_\_ene

How these are named:

- 1) # of C in the ring gives the parent name.
- C-atoms of the double bond are carbon #1 and carbon #2.
- 3) Number through the double bond so that substituents give the lowest possible sum.

Example:

Alkynes: \_\_\_yne

How these are named:

- 1) The longest C-chain with the triple bond is the main chain.
- 2) Number so that the triple bond gets the lowest possible number.

Example:

Benzene-Derived Compounds: \_\_\_benzene

How these are named:

1) Number so that substituents give the lowest possible sum.

Example:

Some benzene derived compounds have unique names. It is good to be familiar with the names and structural formulas of the compounds below, but you do not have to memorize these:

phenol benzaldehyde





benzoic acid

andine

styrene

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