

Chemistry Spring Final Study Guide

Unit 10: Acids & Bases

Use the following information and complete the properties of Acids & Bases.

Acids	Bases
1. Taste	1. Taste
2. Feel	2. Feel
3. Litmus paper	3. Litmus paper
4. pH	4. pH
5. Conductivity	5. Conductivity

Unit 11: Solutions

Draw a line on the graph and provide a description for relationship between Temperature, Volume, and Pressure below.

1. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

2. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

3. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

4. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

5. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

6. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

7. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

8. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

9. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

10. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

11. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

12. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

13. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

14. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

15. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

16. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

17. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

18. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

19. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

20. Draw a line. Describe. Would your volume be 2 percent after the density. Make sure to label your work. There are 20 points at 40%.

Name: _____

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Unit 1: Measurement

Rank the four objects from most accurate to least accurate.

Rank the four objects from most precise to least precise.

Determine the number of Significant Figures in each of the following numbers.

1. 12.3456789 → _____ sig figs

2. 12.3456789 → _____ sig figs

3. 12.3456789 → _____ sig figs

4. 12.3456789 → _____ sig figs

Convert the following into Scientific Notation.

1. 12.3456789 → _____

2. 12.3456789 → _____

3. 12.3456789 → _____

4. 12.3456789 → _____

Convert the following into Decimal Notation.

1. 12.3456789 → _____

2. 12.3456789 → _____

3. 12.3456789 → _____

4. 12.3456789 → _____

Unit 2: Matter

For the following compounds, calculate the **Percent Mass** and the **Empirical Formula** of each compound.

1. H_2O

2. CO_2

3. CH_4

4. SO_2

5. NO_2

6. $FeCl_3$

7. Al_2O_3

8. $Ca(OH)_2$

9. $MgSO_4$

10. KNO_3

11. $NaCl$

12. H_2SO_4

13. $NaOH$

14. $CaCO_3$

15. Fe_2O_3

16. $Al_2(SO_4)_3$

17. Na_2CO_3

18. K_2SO_4

19. $Fe(NO_3)_3$

20. Na_2SO_4

21. $CaSO_4$

22. $Fe_2(SO_4)_3$

23. $Al_2(SO_4)_3$

24. Na_2CO_3

25. K_2SO_4

26. $Fe(NO_3)_3$

27. Na_2SO_4

28. $CaSO_4$

29. $Fe_2(SO_4)_3$

30. $Al_2(SO_4)_3$

31. Na_2CO_3

32. K_2SO_4

33. $Fe(NO_3)_3$

34. Na_2SO_4

35. $CaSO_4$

36. $Fe_2(SO_4)_3$

37. $Al_2(SO_4)_3$

38. Na_2CO_3

39. K_2SO_4

40. $Fe(NO_3)_3$

41. Na_2SO_4

42. $CaSO_4$

43. $Fe_2(SO_4)_3$

44. $Al_2(SO_4)_3$

45. Na_2CO_3

46. K_2SO_4

47. $Fe(NO_3)_3$

48. Na_2SO_4

49. $CaSO_4$

50. $Fe_2(SO_4)_3$

51. $Al_2(SO_4)_3$

52. Na_2CO_3

53. K_2SO_4

54. $Fe(NO_3)_3$

55. Na_2SO_4

56. $CaSO_4$

57. $Fe_2(SO_4)_3$

58. $Al_2(SO_4)_3$

59. Na_2CO_3

60. K_2SO_4

61. $Fe(NO_3)_3$

62. Na_2SO_4

63. $CaSO_4$

64. $Fe_2(SO_4)_3$

65. $Al_2(SO_4)_3$

66. Na_2CO_3

67. K_2SO_4

68. $Fe(NO_3)_3$

69. Na_2SO_4

70. $CaSO_4$

71. $Fe_2(SO_4)_3$

72. $Al_2(SO_4)_3$

73. Na_2CO_3

74. K_2SO_4

75. $Fe(NO_3)_3$

76. Na_2SO_4

77. $CaSO_4$

78. $Fe_2(SO_4)_3$

79. $Al_2(SO_4)_3$

80. Na_2CO_3

81. K_2SO_4

82. $Fe(NO_3)_3$

83. Na_2SO_4

84. $CaSO_4$

85. $Fe_2(SO_4)_3$

86. $Al_2(SO_4)_3$

87. Na_2CO_3

88. K_2SO_4

89. $Fe(NO_3)_3$

90. Na_2SO_4

91. $CaSO_4$

92. $Fe_2(SO_4)_3$

93. $Al_2(SO_4)_3$

94. Na_2CO_3

95. K_2SO_4

96. $Fe(NO_3)_3$

97. Na_2SO_4

98. $CaSO_4$

99. $Fe_2(SO_4)_3$

100. $Al_2(SO_4)_3$

101. Na_2CO_3

102. K_2SO_4

103. $Fe(NO_3)_3$

104. Na_2SO_4

105. $CaSO_4$

106. $Fe_2(SO_4)_3$

107. $Al_2(SO_4)_3$

108. Na_2CO_3

109. K_2SO_4

110. $Fe(NO_3)_3$

111. Na_2SO_4

112. $CaSO_4$

113. $Fe_2(SO_4)_3$

114. $Al_2(SO_4)_$

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DJ Losen



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