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For regular students (except technical stream)

Candidates are required to give their answers in their own words as far as

pro	cticable. The figur	es in the m	argin	indicate full	marks.	
Time: 3 hrs.				Full Marks: 75		
Att	empt all the questio	ns.				
	•	Grou	p 'A'		11×1=11	
	Rewrite the correct	et options o	f each	questions in	your answer sheet.	
1.	2R-OH+2Na → 2RONa+H ₂ , is an example of A) acidic nature of alcohol B) basic nature of alcohol C) electrophilic substitution reaction D) nucleophilic substitution reaction					
2.	C ₆ H ₅ CH ₂ -CHO and C ₆ H ₅ CHO can be distinguished by; A) Iodoform test B) Tollen's test C) 2,4 -DNP test D) Fehling test					
3.	An organic compound (X) undergoes reduction with LiAlH ₄ to yield (Y). When vapour of y are passed over freshly reduced copper at 300°C (X) is formed. The Compound Y is, A) CH ₃ CHO B) CH ₃ CH ₂ -OH C) CH ₃ -CO-CH ₃ D) CH ₃ -O-CH ₃					
4.		B) 2	ral isor		nes of molecular formula D) 4	
5.	Acetic anhydride is obtained from acetyl chloride by the reaction of A) Conc.H,SO ₄ B) P ₂ O ₅ C) CH ₃ COONa C) CH ₃ COOH					
6.	A metal (M) forms thiosulphate with molecular formula M ₂ S ₂ O ₃ . The valency of Metal is					
_	A) 1	B) 2		C) 3	D) 4	
7.	The P ^{II} of 10 ⁻⁸ M a A) less than 7	B) 7	ution o	of HCl is C) 8	D) more than 8	

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Shantanu Bhattacharya, Sanjay Kumar, Avinash K Agarwal

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