



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Their Formulas
(Show the steps for naming the compound)



Steps for Naming:

1. Name the cation (metal) first.
2. Name the anion (non-metal) second, changing the ending to -ide.
3. If the compound is a binary compound, use the prefix system to indicate the number of atoms of each element.
4. If the compound is a polyatomic ion, use the name of the ion.
5. If the compound is an acid, use the name of the acid.
6. If the compound is a base, use the name of the base.

Binary Compounds

First Element	Anion
H ₂	Hydrogen
C	Carbon
N	Nitrogen
O	Oxygen
F	Fluorine
Cl	Chlorine
S	Sulfur
Br	Bromine
I	Iodine
Al	Aluminum
Zn	Zinc
Fe	Iron
Cu	Copper
Ag	Silver
Au	Gold
Pb	Lead

Common Polyatomic Ions

Formula	Name
OH ⁻	Hydroxide
NO ₃ ⁻	Nitrate
SO ₄ ²⁻	Sulfate
PO ₄ ³⁻	Phosphate
CO ₃ ²⁻	Carbonate
ClO ₄ ⁻	Perchlorate
ClO ₃ ⁻	Chlorate
ClO ₂ ⁻	Chlorite
ClO ⁻	Hypochlorite
NO ₂ ⁻	Nitrite
SO ₃ ²⁻	Sulfite
PO ₃ ³⁻	Phosphite

Naming Compounds with a Central and Monatomic



Carbon Dioxide

Steps for Naming:

1. Name the central element first.
2. Name the monatomic anion second, changing the ending to -ide.
3. If the compound is a binary compound, use the prefix system to indicate the number of atoms of each element.

Prefixes:

Prefix	Number	Prefix	Number
mono	1	deca	10
di	2	hepta	7
tri	3	hexa	6
tetra	4	penta	5
penta	5	quad	4
hexa	6	tri	3
hepta	7	di	2
octa	8	mono	1

Common Names for Specific Compounds

Formula	Common Name	Formula	Common Name
H ₂ O	Water	NaCl	Salt
H ₂ O ₂	Hydrogen Peroxide	CaCO ₃	Calcium Carbonate
CH ₄	Methane	NaOH	Sodium Hydroxide
CO ₂	Carbon Dioxide	HCl	Hydrochloric Acid

Chemistry Naming Compounds Study Guide

Leo J. Malone



Chemistry Naming Compounds Study Guide:

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

Study Guide and Solutions Manual Neil E. Schore, 2002-08-02 *Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual* Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter

Organic Chemistry Study Guide with Solutions Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer

Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the

errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2021-09-07 provides a complete guide to the fundamentals of chemistry Page 4 of cover

Study Guide for Chemistry, Third Edition [by] Steven S. Zumdahl Paul B. Kelter, 1993 *Organic Chemistry, Student Study Guide and Solutions Manual* David R. Klein, 2017-01-04 This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry 3e Organic Chemistry 3rd Edition is not merely a compilation of principles but rather it is a disciplined method of thought and analysis Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems Readers must learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Existing textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve problems

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

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