



Name _____

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Wildcat Chemistry

Composition of Drywall: $\text{CaSO}_4 \cdot ? \text{H}_2\text{O}$



Introduction

Hydrates are compounds that incorporate water molecules in their crystalline structures. The ratio of moles of water to one mole of the compound is a small whole number. For example, in the hydrated compound of Epsom salt, $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, the ratio of moles of ionic compound to moles of water is 1:7. This ratio of compound to water can be determined experimentally by heating the hydrate to drive off the water.

The water in a hydrate is bound loosely, and so is relatively easily removed by heating. Most hydrates lose their water of hydration at temperatures slightly above 100°C .



Objective

Find out how many water molecules per hydrated crystal of drywall compound.

Prelab Questions:

Define: Hydrate

Define: Anhydrate

Define: Water of hydration

Procedure

1. Immediately heat your crucible strongly for 3 minutes. Cool and mass.
2. Repeat step 1 and compare the masses. If they are about the same, then you are ready to use the crucible.
3. Remove the paper from the drywall and then grind the sheetrock in a mortar and pestle.
4. Place a 3 to 5 gram sample of the ground sheetrock from the drywall into a crucible.
5. Heat the sample with a fairly strong flame until you see a color change of white to grey. This may take about 5 minutes.
6. After your sample has cooled, and you have massed it, dispose of it in the trashcan.
7. Rinse out the cooled crucible when you are done.

Note: Do NOT place hot crucibles on the balance! Crucibles must be cool enough to touch to place on the balance.

Calculations

Show your calculations **EXACTLY** like the Pre-Laboratory Assignment. Determine what the water of hydration is for drywall compound.

Estimated Formula

Your final statement should be: Therefore, the formula for drywall is $\text{CaSO}_4 \cdot \underline{\hspace{1cm}} \text{H}_2\text{O}$ with the blank filled in.

Lab Format:	
I.	Purpose
II.	PreLab Questions
III.	Procedure
IV.	Data Table
V.	Calculations
VI.	Estimated Formula

Chemistry Lab 11 Composition Of Hydrates Answers

R Sanford



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