

CHEMISTRY MID-TERM STUDY GUIDE --answers

For questions 1-6, determine the number of significant figures in each measurement.

- ___4___1) 4658 g ___2___3) 0.0020 km ___1___5) 200 m
___4___2) 360.0 cm ___3___4) 7.02×10^5 mm ___4___6) 8050. S

For questions 7-10, determine the number of significant figures in the calculated answers without solving the problems.

- ___3___7) $4.033 \times 8.99 =$ ___3___9) $34.6 + 22.22 =$
___2___8) $0.330 / 5.6 =$ ___5___10) $22899 - 2 =$
___d___11) In the measurement 305.67g, which digit is estimated?

a. The 3 b. The 0 c. The 6 d. The 7

___d___12) A piece of paper measures 220.5 cm by 76.3 cm. What is the area of this piece of paper expressed to the correct number of sig. figs.?

a. 16824.15 cm² b. 16824 cm² c. 16820 cm² d. 16800 cm²

___a___13) A sample of metal is found to have a density of 8.44 g/mL and a mass of 10.00g. What is the volume of this metal?

a. 1.18 mL b. 1.180 mL c. 0.844 mL d. 0.84 mL

___b___14) Calculate the density of a gas that has a volume of 1.033×10^{-2} L and a mass of 4.22×10^{-4} g.

a. 4.09×10^2 g/L b. 4.09×10^{-2} g/L c. 2.45×10^3 L/g

For questions 15 - 18 choose "A" if the statement refers to an ionic compound and "B" if the statement refers to a covalent compound.

___B___15) Exist as neutral molecules.

___A___16) Made from a metallic and a nonmetallic element.

___B___17) Made from 2 or more nonmetallic elements.

___A___18) A crystal lattice structure with a very high melting point.

___c___19) How many dots does nitrogen have in its electron dot diagram?

a. 2 b. 4 c. 5 d. 7

Chemistry Honors Midterm Study Guide Answers

D Keegan



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