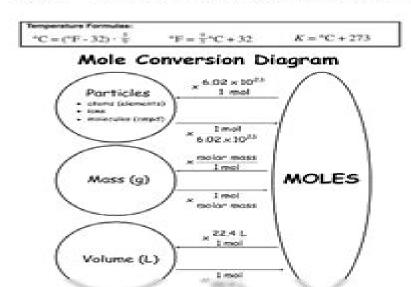
Honors Chemistry B- Final Exam Formula Sheet

Note: You will not use all of these formulas for this test



1 alm + 700 erro Hg + 101.3 kPu P = v - P = v .

P___ + P_ + P_ + P_ + ...

Service Company

Processe Na/2 - Married Married - 1987s.

$$M_1 \times V_1 = M_2 \times V_2$$

Percent by Mass= Mass of Solution ×100

Percent by Volume = Volume of Solution x100

Mole fractions: $X_A = \frac{n_A}{n_A + n_B}$ $X_B = \frac{n_B}{n_A + n_B}$

1 Calorie = 1 kilocalorie = 1000 calories

$$Q = mc\Delta T$$

$$q_{\text{sys}} = \Delta H = -q_{\text{surr}} = -\text{m} \times C \times \Delta T$$

$$\Delta H^0 = \Delta H^0$$
(products) - ΔH^0 (reactants)

 $\Delta G = \Delta N - T\Delta S$

radio - ACAL TIME

piti = -kog(H)

pOH = -logfOH

] = [OH] = 1.0 = 10 14

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Mark Steyvers

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