Key

	Electron Configuration Evaluation
	space below, write the unabbreviated electron configurations of the ing elements:
13	sodium /5 2 25 2 p 35 2
2)	iron 152 252 2p6 352 3p6 452 3d
3)	bromine 152 252 2p6 3523p6 4523d104p
4)	barium 152 252 206 352 306 452 3010 4p 5524 10 57 652
5)	neptunium 15226265353545236452510496562461650665246145d1060
In the	space below, write the abbreviated electron configurations of the following ents:
6)	cobalt LAr 45 ² 3d
7)	silver LKrJ 5524d 9
8)	tellurium [Kr] 552 4010 5p4
9)	radium IRn 752
10)	lawrencium [Rn] 752 5F /4
Dete	ermine what elements are denoted by the following electron configurations:
11)	1s ² 2s ² 2p ⁶ 3s ² 3p ⁴ 5
12)	1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 4s ² 3d ¹⁰ 4p ⁶ 5s ¹ Rb
13)	[Kr] 5s ² 4d ¹⁰ 5p ³ A.S
14)	[Xe] 6s ² 4f ¹⁴ 5d ⁶
15)	[Rn] 7s ² 6d'5f''
Det	ermine which of the following electron configurations are not valid:
6	1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 4s 4d ¹ p ⁵ 3r
(17)	1s ² 2s ² 2p 3s ³ 3d mn
(18)	(Ra)7s2518BK
19)	[Kr] 5s ² 4d ¹⁰ 5p ⁵
20)	[Xe] Xe

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