

FINAL EXAM TOPICS

The exam will consist of 40 multiple choice questions, 12 from exam 1, 12 from exam 2, 12 from exam 3 and 4 from the material covered since exam 3. The latter material corresponds to Chap. 12, section 7 and Chap. 21, sections 1-4 (you're responsible for material covered in lecture only). As usual the exam will consist of 70% "C" questions, 20% "B" questions, and 10% "A" questions. I will determine what are A, B and C questions from the midterm exams by considering how many people answered the question correctly on the exam.

"C" Questions on New Material(Homework Examples)

Chapter 12:

- Be able to construct and/or interpret a simple reaction coordinate diagram for a multi-step reaction and understand how it relates to the mechanism.
- Know what a reaction mechanism is, and the difference between an elementary step and the overall reaction.
- Write rate laws for elementary reactions. [12.73]
- Know what the significance of the slow or "rate-determining step" in a reaction is.
- Be able to determine the overall reaction from a series of elementary reactions.
- Be able to identify reaction intermediates in a mechanism.
- Know what a catalyst is and how it affects (and does not affect) a chemical reaction. [12.77]
- Be able to identify the catalyst and the catalytic intermediate in a mechanism.[12.79]
- Be able to identify the catalytic pathway in a transition state diagram for a reaction. [12.80 (b,b), 12.81]

Chapter 21:

- Know what alpha emission, beta emission, electron capture, positron emission are. Be able to balance nuclear reactions. [worksheet 4, 21.11, 21.21, 21.13, 21.45]
- Know what half-life and the decay constant are. Be able to convert back and forth between the two.[21.37]
- Use the integrated rate law for first order processes to determine time info given half life and some information about the amounts of radioactive material. [21.34b (21 min)]
- Use the integrated rate law for first order processes to determine info about amounts of radioactive material given half life and time. [21.32(2.3%, 2.0×10^{-11} %), 21.33]

"B" Questions on New Material (Homework Examples)

Chapter 12:

- Decide whether a given reaction mechanism is reasonable based on the simple criteria outlined in class, and, in particular, whether it is consistent with the experimental rate law.

Chapter 21:

- Calculate the energy changes involved in nuclear reactions from the mass defects. [21.43, 21.54 (2.0×10^8 kJ/mol)]

"A" Questions (Homework Examples)

The other topics listed under B and C questions can be converted to A questions. In general, "A" questions are those which require bringing together different concepts and where the route to solve the problem is not implicitly given.

Material given with the exam: Periodic Table; Table of K_a and K_b values; Table of E° values; All the equations given in the midterm exams; nuclear chemistry equations.

Chemistry Final Exam Study Guide 201

Julia Schneider



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